Direct cupration of fluoroform:

A novel and economical route to CuCF₃ for a variety of trifluoromethylation reactions

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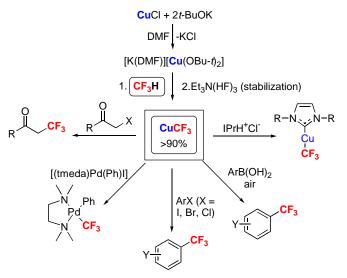
Direct cupration of fluoroform: A novel and economical route to CuCF₃ for a variety of trifluoromethylation reactions

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Trifluoromethylated building blocks and intermediates are in exceptionally high demand for the synthesis of agrochemicals, pharmaceuticals, and specialty materials. Readily available fluoroform, CHF₃, a side-product of Teflon manufacturing, would be by far the best CF₃ source for various trifluoromethylation reactions.¹ Chemoselective activation of fluoroform, however, is highly challenging.

We have recently discovered the first reaction of direct cupration of fluoroform² and established its mechanism.³ Treatment of CuCl reacts with 2 equiv of *t*-BuOM (M = K, Na) in DMF produces novel dialkoxycuprates $[M(DMF)_n]^+[Cu(OBu-t)_2]^-$ (X-ray) that readily metalate CHF₃ at room temperature and atmospheric pressure.² The resultant CuCF₃ (>90% yield) has been successfully used for highly efficient,



low-cost trifluoromethylation reactions of a variety of organic and inorganic substrates.^{2,4-7}

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- (7) Lishchynskyi, A.; Grushin, V. V. J. Am. Chem. Soc. 2013, 135, 12584.



Vladimir Grushin, a native of Moscow, Russia, obtained his PhD degree from Moscow State University (1984). He then spent several years doing research at the Institute of Organo-Element Compounds of the Russian Academy of Sciences and at the University of Ottawa before joining the faculty at Wilfrid Laurier University, Ontario, Canada. In 1997, he took a research position at DuPont CR&D in Wilmington, Delaware. After 12 years with DuPont, he returned to academia in 2010 with an appointment as Senior Group Leader at the Institute of Chemical Research of Catalonia (ICIQ) in Tarragona, Spain. His research interests span organic and inorganic chemistry, including catalysis and organometallic fluorine chemistry.

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